# Course description

Course abbreviation:	KFCH/C120 Equilibrium T	hermodynamics				Page:	1 / 2	
Academic Year:	2023/2024	nermodynamics			Printed:	29.05.2024	09:22	
Department/Unit /	KFCH / C120			Academic Year	2023/2024			
Title	Equilibrium T	hermodynamics		Type of completion	Examination			
Accredited/Credits	Yes, 5 Cred.				Type of completion	Combined		
Number of hours	Lecture 2 [HRS/WEEK] Seminar 2 [HRS/WEEK]							
Occ/max	Status A	Status B	Status C		Course credit prior to	NO		
Summer semester	0 / -	0 / -	0 / -		Counted into average	YES		
Winter semester	2 / -	0 / -	0 / -		Min. (B+C) students	not determ	ined	
Timetable	Yes				Repeated registration	NO		
Language of instruction	Czech				Semester taught	Winter sen	nester	
Optional course	Yes				Internship duration	0		
Evaluation scale	A B C D E F							
No. of hours of on-premise								
Auto acc. of credit	No							
Periodicity	К							
Substituted course	None							
Preclusive courses	N/A							
Prerequisite courses	N/A							
Informally recommended courses		N/A						
Courses depending on this Course		N/A						

## Course objectives:

The course brings information about behavior of the substances in external force fields and about of estimation of properties of pure substances and their mixtures using the advanced equations of state and information about theory of activity coefficient.

#### Requirements on student

Regular weekly checking of the knowledge by solving theoretical and numerical problems - 40% of the subject examination Final examination (oral knowledge verification) - 60% of the subject examination

#### Content

Using more sophisticated equations of state for the determination of thermodynamics properties of real substances. Thermodynamics of systems in the gravitation, centrifugal, electrostatic and magnetic fields. Thermodynamics of low temperatures - cryogenics. Theory of solutions - activity coefficients.

#### Prerequisites - other information about course preconditions

Knowledge on the level of the Advanced Physical Chemistry course

#### Competences acquired

The course enhances knowledge of Chemical Thermodynamics compared to course Advanced Physical Chemistry

## Fields of study

# Guarantors and lecturers

• Guarantors: doc. Ing. Pavel Čičmanec, Ph.D. (100%)

- Lecturer: doc. Ing. Pavel Čičmanec, Ph.D. (100%)
- Seminar lecturer: doc. Ing. Pavel Čičmanec, Ph.D. (100%)

# Literature

• Basic:	Novák J., Růžička K. Chemická termodynamika I. VŠCHT Praha, 2002.
• Basic:	Dohnal V., Novák J., Matouš J. Chemická termodynamika II. VŠCHT Praha, 1997.
• Recommended:	SANDLER S.I. Chemical and Engineering Thermodynamics, 3th Ed John Wiley & Sons, New York, 1999.

# Teaching methods

Work with text (with textbook, with book)

### Assessment methods

Work-related product analysis

# Course is included in study programmes:

Study Programme	Type of	Form of	Branch	Stage St.	plan v.	Year	Block	Status	R.year	R.
Physical Chemistry	Follow-up study	Full-time	Physical Chemistry	1 2	2022	2023	povinné předměty	А	1	ZS
Physical Chemistry	Follow-up study	Full-time	Physical Chemistry	1 2	2023	2023	povinné předměty	А	1	ZS
Inorganic and Bioinorganic Chemistry	Follow-up study	Full-time	Inorganic and Bioinorgan Chemistry	ic 12	2022	2023	volitelné předměty	С	1	ZS
Inorganic and Bioinorganic Chemistry	Follow-up study	Full-time	Inorganic and Bioinorgan Chemistry	ic 12	2023	2023	volitelné předměty	С	1	ZS